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Code No. : S-253

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Annual Examination - 2018

B.Sc. Part - II

PHYSICS

Paper - I

THERMODYNAMICS, KINETIC THEORY AND

STATISTICAL PHYSICS

Max.Marks : 50

Min.Marks : 17

Time : 3 Hrs.

Vhi % [k.M ^v* eanl vfry?kjkjh izu gftlgagy djuk vfuok; ZgA [k.M ^c* eay?kjkjh c'u , oa [k.M ^1 * eanh?kjkjh c'u gA [k.M ^v* dks l cl sigysgy djA

Note : Section 'A', containing 10 very short-answer-type questions, is compulsory. Section 'B' consists of short-answer-type questions and Section 'C' consists of long-answer-type questions. Section 'A' has to be solved first.

Section - 'A'

fuEukdr vfry?kjkjh c'ula ds mYkj , d ; k nls okD; ka ea na Answer the following very short-answer-type questions in one or two sentences. (1x10=10)

c'u 1- , .vRWh dks i fjHkkf"kr dhft , A

Define entropy.

c'u 2- C_p , oa C_v ea D; k l eak gS

What is the relation between C_p and C_v?

c'u 3- bat u dh n{krk l sD; k rkr i ; ZgS

What do you mean by efficiency of an engine?

c'u 4- , .vRWh , oa A"ekxfrd i kf; drk ea l eak nhft , A

Give the relation between entropy and thermodynamic probability.

P.T.O.

c'u 2- t_w-FkkE l u i Hkko l svki D; k l e>rsgA n'kkb, fd vkn'kz xS dsfy, t_w-FkkE l u xqkkad] μ = 0 gkrk gA

What do you mean by Joule-Thomson effect? Show that for an ideal gas Joule-Thomson coefficient, μ = 0 .

OR

N".k fi .M fofdj.k dsfy, fl) dhft, λ_T = fu; rkd

For Black-body radiation, prove that λ_T = constant.

c'u 3- fdl h xS ds'; kurk xqkkad dsfy, 0; atd fuxfer dhft , A

Deduce an expression for the coefficient of viscosity of a gas.

OR

ghfy; e ds aohdj.k dh fof/k dk o.ku dhft , A

Describe a method to liquify helium.

c'u 4- ckVtEeS dSukudy fu; e fyf[k, rFkk fl) dhft , A

State and prove Boltzmann canonical law.

OR

, d foeh; ckDI ea cln d.k dh A'tk dsfy, 0; atd iklr dhft , A

Obtain an expression for the energy of a particle enclosed in a one dimensional box.

c'u 5- Qehz fMjkd l ka[; dh dsfy, forj.k Qyu iklr dhft , A

Obtain distribution function for Fermi-Dirac statistics.

OR

eDI oy-ckVtEeS l ka[; dh] ckl -vkbUI Vhu , oa Qehz fMjkd l ka[; dh ea vlrj Li"V dhft , A

Distinguish between Maxwell-Boltzmann statistics, Bose-Einstein statistics, and Fermi-Dirac statistics.

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ç'u 5- fofdj.k dsfy, fdjpkQ dk fu; e fyf[k, A

State Kirchhoff's law for radiation.

ç'u 6- I k_s fu; rka dks ifjHkkf"kr dhft, A

Define solar constant.

ç'u 7- i kf; drk dks ifjHkkf"kr dhft, A

Define probability.

ç'u 8- dyk vkdk'k I sD; k rkrri ; ZgS

What do you mean by phase space?

ç'u 9- Qehz/kkU D; k gS

What is fermion?

ç'u 10- Qehz Åtkz dks ifjHkkf"kr dhft, A

Define fermi energy.

Section - 'B'

fuEukdr y?lq mYkj; ç'ula ds mYkj 150&200 'kn I hek ea na
Answer the following short-answer-type questions with word
limit 150-200 (3x5=15)

ç'u 1- Å"ekxfrdh dk 'kk; okafu; e fyf[k, rFkk rki dh vfhk/kkj.kk dks I e>kb, A

Write the Zeroth law of thermodynamics and explain the concept of temperature.

OR

ekxZ Qyu , ofclnq Qyu ea vlrj Li"V dhft, A

Distinguish between path function and point function.

ç'u 2- xlr Å"ek dk Dykfl ; I Dys jkkU I ehdj.k 0; qilu dhft, A

Establish Clausius-Clapeyron's equation of latent heat.

OR

fofdj.k I ckh lykd dh Dok/e ifjdYi uk dks fyf[k, A

Write Planck's quantum hypothesis of radiation.

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ç'u 3- xS ka ea vfhkxeu ?kVukvka dks I e>kb, A

Explain the transport phenomenon in gases.

OR

o.kØe j[kkvka dk Mklyj folrhdj.k dks I e>kb, A

Explain the Doppler broadening of spectral lines.

ç'u 4- vfhkxE; I Ie voLFkvka dks mnkgj.k I fgr I e>kb, A

Explain accessible microstates with example.

OR

I kf[; dh ; kf=dh dh eq[; vfhkdYi uk, j fyf[k, A

Write the fundamental postulates of statistical mechanics.

ç'u 5- Qehz fMjkd I kf[; dh dh emy vfhkdYi ukvka dks fyf[k, A

State the basic assumptions of Fermi-Dirac statistics.

OR

folks| d.k rFkk vfolks| d.k ea vlrj Li"V dhft, A

Distinguish between distinguishable and indistinguishable particles.

Section - 'C'

fuEukdr nhz mYkj; ç'ula ds mYkj 300&350 'kn I hek ea na
Answer the following long-answer-type questions with word
limit 300-350 (5x5=25)

ç'u 1- Å"ekxfrdh ds f}rh; fu; e ds nksuka dFkuka dks fyf[k, rFkk mudh I erY; rk n'kkb, A

State both the statements of Second law of thermodynamics and show their equivalence.

OR

dsYou dsrki ds Å"ekxfrdh eki Øe dks I e>kb, , oabl eki Øe ds 'kk; rFkk fMxh ds vkdkj dks I e>kb, A

Explain Kelvin's thermodynamic scale of temperature and explain zero of the scale and size of the degree.

P.T.O.